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Scope and Limitations of Functionalized Acetylenic Triflones in the Direct Alkynylation of C-H Bonds.¹

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Abstract: Reactions of TIPS-substituted acetylenic triflones extends the versatility of the trifluoromethyl radical mediated C-H alkynylation reaction. Alkynes bearing acetylenic or propargylic oxygen functionality cannot be prepared, but silyl ethers or chlorides in remote positions may be carried through the reaction. Crown ethers are successfully monofunctionalized and distal bis-acetylenes may be employed in the title reaction. © 1997 Elsevier Science Ltd.

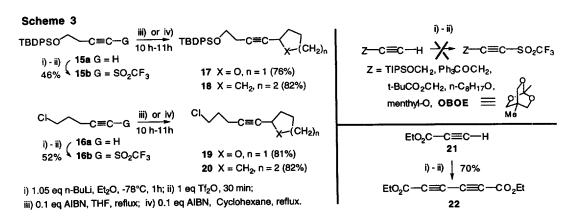
We recently reported that reaction of ethers, sulfides, and hydrocarbons with acetylenic triflones such as 2A,B provides facile access to substituted alkynes 3 (Scheme 1).² The reaction proceeds via radical C-H abstraction by the very electrophilic trifluoromethyl radical 4 in a process involving addition of alkyl radical 5 to the α -carbon of the acetylenic triflone 2 followed by elimination of vinyl radical 6 to alkyne 3 and trifluoromethylsulfonyl radical 7. Fragmentation of 7 to sulfur dioxide and trifluoromethyl radical 4 propagates the chain.

In order to delineate the scope and limitations of reagents bearing additional functionality we undertook the synthesis of acetylenic triflones bearing a silane. The exceptional electron-withdrawing ability of the triflone imparts considerable lability to the β-silylacetylenic triflones, attempts at synthesis of TMS-substituted reagent 9a leading to black tars. TBDMS-alkynyl triflone 9b was also somewhat sensitive, but could be isolated and employed in the alkynylation reaction, affording THF adduct 10b in a moderate yield of 63%. The TIPS-substituted reagent 9c was far more satisfying, providing excellent yields of C-H alkynylated products 10c-13c (Scheme 2). As previously observed,² reactions conducted in acetonitrile solution are somewhat slower and lower-

yielding than those run in neat substrate. For example, adamantane is converted to acetylenic silane **14c** in 50% yield after heating in acetonitrile with reagent **8c** for 48h (Scheme 2).

Scheme 2 1a. n-BuLi Et₂O, -78°C
$$Si-C = C-SO_2CF_3$$
 refl. 4-48h, 15% AIBN (CH₂) (

Distal substituents such as homopropargyl ether 15b and bishomopropargyl chloride 16b undergo the alkynylation reaction with test substrates tetrahydrofuran and cyclohexane affording the adducts 17-20 in 76-82% yield (Scheme 3, left). Unfortunately all attempts with substrates bearing functionality in closer proximity to the alkyne moiety, such as protected propargyl alcohols, acetylenic ethers, or the indicated bicyclooctyl orthoester (OBOE³), were uniformly disastrous. While the acetylenic anions could be prepared in each instance, addition of triflic anhydride generated a reaction mixture which began to darken even at low temperature and upon further warming only produced intractable tarry mixtures (Scheme 3, top right). The only substrate which yielded an identifiable product was ethyl propiolate 21, which afforded dimer 22, via additionelimination of the acetylenic anion to the incipient acetylenic triflone (Scheme 3, bottom right).



Acetylenic triflones **2A** and **9c** are also excellent reagents for the alkynylation of crown ethers. As can be seen in Scheme 4 and Table 1, both 12-crown-4 **23** and 15-crown-5 **24** were effectively monoalkynylated to provide functionalized crowns **25-26**. Desilylation of **25c** and **26c** could be effected in near-quantitative yield (1.1equiv TBAF, 25°C, 2h). This represents an extremely efficient access to molecules of this nature.⁴

Scheme 4

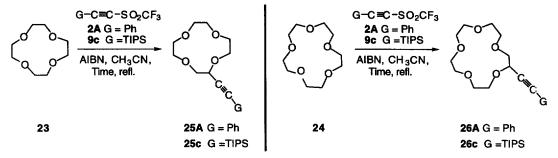


Table 1. Comparison of reactions of Crown ethers with Acetylenic Triflones **2A** and **9c**.

Run	Crown (#equiv)	Conditions#	Products	Yield
1	23 (10eq)	2A, 3h, 0.10M	25A	70%
2	23 (1.5eg)	2A , 48h, 0.05M	25A	55%*
3	23 (5eq)	9c, 6h, 0.06M	25c	70%
4	23 (1.5eq)	9c, 48h, 0.08M	25c	50%
5	24 (10eg)	2A, 2h, 0.10M	26A	80%
6	24 (1.5eg)	2A, 24h, 0.05M	26A	60%*
7	24 (5eq) "	9c, 5h, 0.06M	26c	70%
8	24 (1.5eq)	9c, 48h, 0.05M	26c	53%

#AIBN (20 mol%), acetonitrile at reflux,; *20% PhCOCH₂SO₂CF₃ also isolated.

The final class of reagents investigated were the bis-acetylenes. As might now be expected, the 1,3 and 1,4 diacetylenes were not satisfactory substrates. However the 1,7-diynes shown in Scheme 5 were perfectly well-behaved materials. Although direct monofunctionalization of 27 proceeded in only 30% yield, the alkynylation reactions of the resulting 28 proved that the C-H functionalization proceeds in the presence of the unprotected terminal acetylene (Scheme 5, rxn 1). A higher-yielding, but longer sequence involves starting with TIPS-acetylene 31 and provides 29Si and 30Si, the TIPS-protected derivatives of compounds 29H and 30H. Fluoride-mediated cleavage of these materials then affords 29H and 30H in near-quantitative yield (Scheme 5, rxn 2). Application of the metalation/sulfonylation sequence gives acetylenic triflones 29Tf and 30Tf in reasonable yield (Scheme 5, rxn 3). Reaction of 29Tf with cyclohexane or 30Tf with THF provide the identical bis-functionalized bis-acetylene 34 in >80% yield in each instance. Treatment of 30Tf

with cyclohexane provides the expected symmetrical bis-acetylene 35 in similar yield (Scheme 5, rxn 3).

TIPS-C=C-H
$$\xrightarrow{\text{V}}$$
 TIPS-C=C-(CH₂)₄-C=C-G $\xrightarrow{\text{iii}}$ or TIPS-C=C-(CH₂)₄-C=C- $\xrightarrow{\text{V}}$ TIPS-C=C-(CH₂)₄-C=C- $\xrightarrow{\text{V}}$ $\xrightarrow{\text{V}}$ 29H (98%) 31 $\xrightarrow{\text{I}}$ 32 G = H $\xrightarrow{\text{IV}}$ 32 G = H $\xrightarrow{\text{IV}}$ 33 G = SO₂CF₃ 30Si X = CH₂, n = 2 (82%) (2)

i) 1.05 eq n-BuLi, Et₂O, -78°C, 1h; ii) 1 eq Tf₂O, 30 min; iii) 0.15 eq AlBN, THF, reflux, 14h; iv) 0.15 eq AlBN, Cyclohexane, reflux, 24h; v) n-BuLi -78°C THF/HMPA, add 1,4-dibromobutane 5eq., warm to 23°C (98%); vii) TMS-C=C-Li, warm to 23°C, (97%); viii) K_2 CO₃ 1eq, MeOH, 0°C to 23°C, 3 h (83%); viii) TBAF, THF, 23°C, 2 h, (98%)

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References and Notes

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